



NEET Exam. 2024 (05th May 2024)

Code-T3 (Paper & Answer)

General Instruction : This paper contains **200 questions** out of which candidates need to answer **180 questions**. The questions will be divided into two sections for each subject : **Section-A** and **Section-B**. Section A contains 35 questions while Section B will have 15 questions. **Out of these 15 questions, candidates can answer any 10, which makes it 180 questions.**

Marking Scheme : +4 marks will be awarded for each correct answer & –1 marks deducted for wrong answer.

Subjects	Sections	No. of questions	Section-wise Marks
Physics	Section A	35	140
	Section B	15	40
Chemistry	Section A	35	140
	Section B	15	40
Botany	Section A	35	140
	Section B	15	40
Zoology	Section A	35	140
	Section B	15	40
Total Marks			720

PHYSICS

SECTION-A

Q.1 A tightly wound 100 turns coil of radius 10 cm carries a current of 7 A. The magnitude of the magnetic field at the centre of the coil is (Take permeability of free space as $4\pi \times 10^{-7}$ SI units)
(1) 4.4 mT (2) 44 T (3) 44 mT (4) 4.4 T

Ans. [1]

Q.2 Match List-I with List-II.

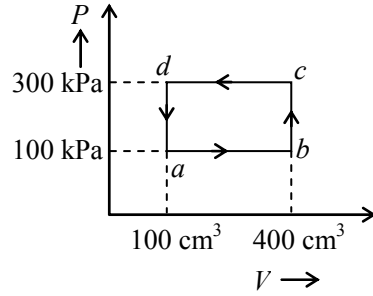
	List-I (Material)		List-II (Susceptibility (χ))
A.	Diamagnetic	I.	$\chi = 0$
B.	Ferromagnetic	II.	$0 > \chi \geq -1$
C.	Paramagnetic	III.	$\chi \gg 1$
D.	Non-magnetic	IV.	$0 < \chi < \varepsilon$ (a small positive number)

Choose the correct answer from the options given below :

- (1) A-III, B-II, C-I, D-IV (2) A-IV, B-III, C-II, D-I
(3) A-II, B-III, C-IV, D-I (4) A-II, B-I, C-III, D-IV

Ans. [3]

Q.3 A thermodynamics system is taken through the cycle *abcd*. The work done by the gas along the path *bc* is-



- (1) -90 J (2) -60 J (3) zero (4) 30 J

Ans. [3]

Q.4 An unpolarised light beam strikes a glass surface at Brewster's angle. Then

- (1) both the reflected and refracted light will be completely polarised
 (2) the reflected light will be completely polarised but the refracted light will be partially polarised
 (3) the reflected light will be partially polarised
 (4) the refracted light will be completely polarised

Ans. [2]

Q.5 In an ideal transformer, the turns ratio is $\frac{N_p}{N_s} = \frac{1}{2}$. The ratio $V_s : V_p$ is equal to (the symbols carry their usual meaning) -

- (1) 1 : 1 (2) 1 : 4 (3) 1 : 2 (4) 2 : 1

Ans. [4]

Q.6 A logic circuit provides the output Y as per the following truth table :

A	B	Y
0	0	1
0	1	0
1	0	1
1	1	0

The expression for the output Y is-

- (1) \bar{B} (2) B (3) $A \cdot B + \bar{A}$ (4) $A \cdot \bar{B} + \bar{A}$

Ans. [1]

Q.7 In a vernier calipers, (N + 1) divisions of vernier scale coincide with N divisions of main scale. If 1 MSD represent 0.1 mm, the vernier constant (in cm) is-

- (1) 100 N (2) 10(N + 1) (3) $\frac{1}{10 N}$ (4) $\frac{1}{10(N+1)}$

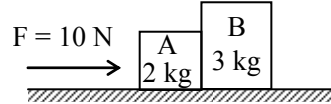
Ans. [4]

Q.8 The maximum elongation of steel wire of 1 m length if the elastic limit of steel and its Young's modulus, respectively, are $8 \times 10^8 \text{ N m}^{-2}$ and $2 \times 10^{11} \text{ N m}^{-2}$, is-

- (1) 40 mm (2) 8 mm (3) 4 mm (4) 0.4 mm

Ans. [3]

Q.9 A horizontal force 10 N is applied to a block A as shown in figure. The mass of blocks A and B are 2 kg and 3 kg, respectively. The blocks slide over a frictionless surface. The force exerted by block A on block B is-



- (1) 6 N (2) 10 N (3) zero (4) 4 N

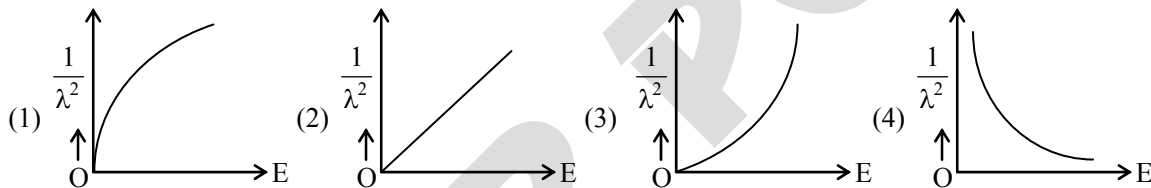
Ans. [1]

Q.10 If the monochromatic source in Young's double slit experiment is replaced by white light, then-

- (1) there will be a central bright white fringe surrounded by few coloured fringes
 (2) all bright fringes will be of equal width
 (3) interference pattern will disappear
 (4) there will be a central dark fringe surrounded by a few coloured fringes

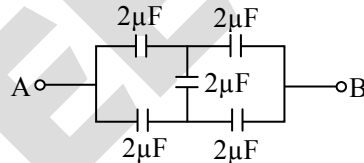
Ans. [1]

Q.11 The graph which shows the variation of $\left(\frac{1}{\lambda^2}\right)$ and its kinetic energy, E is (where λ is de Broglie wavelength of a free particle)



Ans. [2]

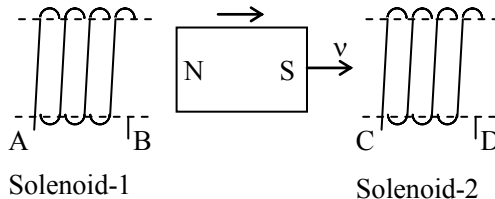
Q.12 In the following circuit, the equivalent capacitance between terminal A and terminal B is-



- (1) 0.5 μF (2) 4 μF (3) 2 μF (4) 1 μF

Ans. [3]

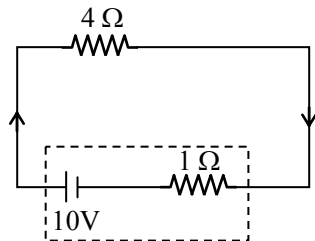
Q.13 In the diagram, a strong bar magnet is moving towards solenoid-2 from solenoid-1. The direction of induced current in solenoid-1 and that in solenoid-2, respectively, are through the directions :



- (1) AB and CD (2) BA and DC (3) AB and DC (4) BA and CD

Ans. [3]

- Q.18** The terminal voltage of the battery, whose emf is 10 V and internal resistance 1 Ω , when connected through an external resistance of 4 Ω as shown in the figure is-



- (1) 8 V (2) 10 V (3) 4 V (4) 6 V

Ans. [1]

- Q.19** Match List I with List II.

	List-I (Spectral Lines of Hydrogen for transitions from)		List-II (Wavelengths (nm))
A.	$n_2 = 3$ to $n_1 = 2$	I.	410.2
B.	$n_2 = 4$ to $n_1 = 2$	II.	434.1
C.	$n_2 = 5$ to $n_1 = 2$	III.	656.3
D.	$n_2 = 6$ to $n_1 = 2$	IV.	486.1

Choose the correct answer from the options given below :

- (1) A-IV, B-III, C-I, D-II (2) A-I, B-II, C-III, D-IV
 (3) A-II, B-I, C-IV, D-III (4) A-III, B-IV, C-II, D-I

Ans. [4]

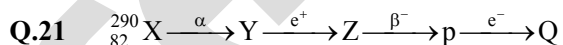
- Q.20** If c is the velocity of light in free space, the correct statements about photon among the following are

- A. The energy of a photon is $E = hv$.
 B. The velocity of a photon is c .
 C. The momentum of a photon, $p = \frac{hv}{c}$.
 D. In a photon-electron collision, both total energy and total momentum are conserved.
 D. Photon possesses positive charge.

Choose the correct answer from the options given below :

- (1) A, C and D only (2) A, B, D and E only
 (3) A and B only (4) A, B, C and D only

Ans. [4]



In the nuclear emission stated above, the mass number and atomic number of the product Q respectively, are:

- (1) 288, 82 (2) 286, 81 (3) 280, 81 (4) 286, 80

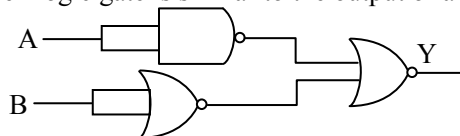
Ans. [2]

- Q.22** At any instant of time t , the displacement of any particle is given by $2t - 1$ (SI unit) under the influence of force of 5 N. The value of instantaneous power is (in SI unit) :

- (1) 7 (2) 6 (3) 10 (4) 5

Ans. [3]

Q.23 The output (Y) of the given logic gate is similar to the output of an/a :



- (1) OR gate (2) AND gate (3) NAND gate (4) NOR gate

Ans. [2]

Q.24 The mass of a planet is $\frac{1}{10}$ th that of the earth and its diameter is half that of earth. The acceleration due to gravity on that planet is-

- (1) 4.9 m s^{-2} (2) 3.92 m s^{-2} (3) 19.6 m s^{-2} (4) 9.8 m s^{-2}

Ans. [2]

Q.25 Given below are two statements :

Statement I : Atoms are electrically neutral as they contain equal number of positive and negative charges.

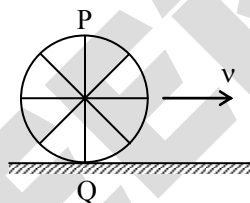
Statement II : Atoms of each element are stable and emit their characteristic spectrum.

In the light of the above statements, choose the most appropriate answer from the options given below.

- (1) Statement I is correct but Statement II is incorrect
 (2) Statement I is incorrect but Statement II is correct
 (3) Both Statement I and Statement II are correct
 (4) Both Statement I and Statement II are incorrect

Ans. [1]

Q.26 A wheel of a bullock cart is rolling on a level road as shown in the figure below. If its linear speed is v in the direction shown, which one of the following options is correct (P and Q are any highest and lowest points on the wheel, respectively) ?



- (1) Both the points P and Q move with equal speed
 (2) Point P has zero speed
 (3) Point P moves slower than point Q
 (4) Point P moves faster than point Q

Ans. [4]

Q.27 A particle moving with uniform speed in a circular path maintains :

- (1) constant velocity but varying acceleration
 (2) varying velocity and varying acceleration
 (3) constant velocity
 (4) constant acceleration

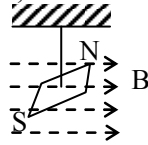
Ans. [2]

Q.28 A thin flat circular disc of radius 4.5 cm is placed gently over the surface of water. If surface tension of water is 0.07 Nm^{-1} , then the excess force required to take it away from the surface is :

- (1) 1.98 mN (2) 99 N (3) 19.8 mN (4) 198 N

Ans. [3]

- Q.29** In a uniform magnetic field of 0.049 T, a magnetic needle performs 20 complete oscillations in 5 seconds as shown. The moment of inertia of the needle is $9.8 \times 10^{-6} \text{ kg m}^2$. If the magnitude of magnetic moment of the needle is $x \times 10^{-5} \text{ Am}^2$; then the value of 'x' is-



- (1) $50 \pi^2$ (2) $1280 \pi^2$ (3) $5 \pi^2$ (4) $128 \pi^2$

Ans. [2]

- Q.30** Two bodies A and B of same mass undergo completely inelastic one dimensional collision. The body A moves with velocity v_1 while body B is at rest before collision. The velocity of the system after collision is v_2 . The ratio $v_1 : v_2$ is-

- (1) 4 : 1 (2) 1 : 4 (3) 1 : 2 (4) 2 : 1

Ans. [4]

- Q.31** If $5 \sin\left(\pi t + \frac{\pi}{3}\right) \text{ m}$ represents the motion of a particle executing simple harmonic motion, the amplitude and time period of motion, respectively, are;

- (1) 5 cm, 1 s (2) 5 m, 1 s (3) 5 cm, 2 s (4) 5 m, 2 s

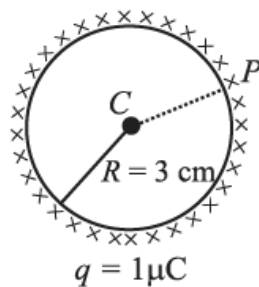
Ans. [4]

- Q.32** The quantities which have the same dimensions as those of solid angle are;

- (1) strain and arc (2) angular speed and stress
(3) strain and angle (4) stress and angle

Ans. [3]

- Q.33** A thin spherical shell is charged by some source. The potential difference between the two points C and P (in V) shown in the figure is (Take $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ SI units}$)



- (1) 0.5×10^5 (2) zero (3) 3×10^5 (4) 1×10^5

Ans. [2]

- Q.34** A bob is whirled in a horizontal plane by means of a string with an initial speed of $\omega \text{ rpm}$. The tension in the string is T. If speed becomes 2ω while keeping the same radius, the tension in the string becomes;

- (1) $\frac{T}{4}$ (2) $\sqrt{2} T$ (3) T (4) 4T

Ans. [4]

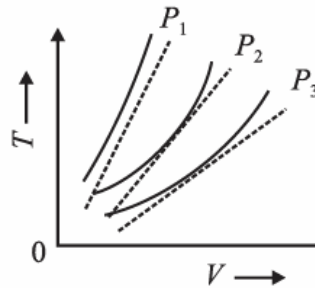
Q.35 A wire of length ' ℓ ' and resistance 100Ω is divided into 10 equal parts. The first 5 parts are connected in series while the next 5 parts are connected in parallel. The two combinations are again connected in series. The resistance of this final combination is;

- (1) 55Ω (2) 60Ω (3) 26Ω (4) 52Ω

Ans. [4]

SECTION-B

Q.36 The following graph represents the T-V curves of an ideal gas (where T is the temperature and V the volume) at three pressures P_1, P_2 and P_3 compared with those of Charles's law represented as dotted lines.



Then correct relation is:

- (1) $P_2 > P_1 > P_3$ (2) $P_1 > P_2 > P_3$ (3) $P_3 > P_2 > P_1$ (4) $P_1 > P_3 > P_2$

Ans. [2]

Q.37 A parallel plate capacitor is charged by connecting it to a battery through a resistor. If I is the current in the circuit, then in the gap between the plates;

- (1) displacement current of magnitude equal to I flows in a direction opposite to that of I.
 (2) displacement current of magnitude greater than I flows but can be in any direction.
 (3) there is no current.
 (4) displacement current of magnitude equal to I flows in the same direction as I.

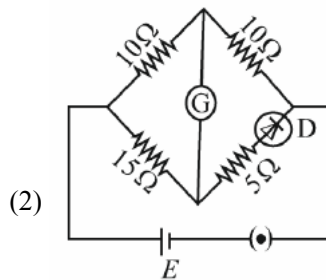
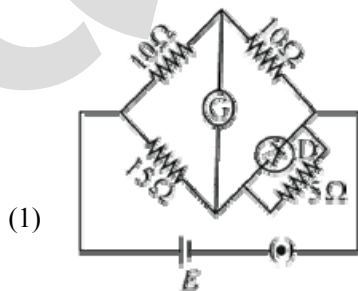
Ans. [4]

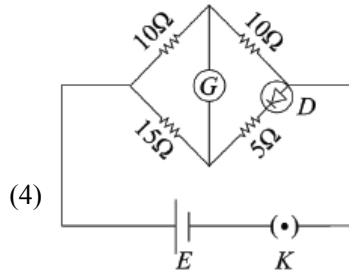
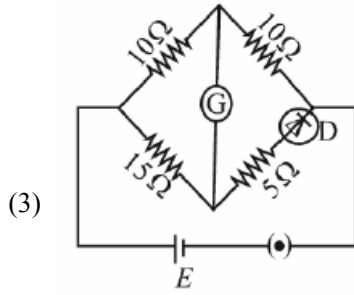
Q.38 The property which is not of an electromagnetic wave travelling in free space is that;

- (1) they travel with a speed equal to $\frac{1}{\sqrt{\mu_0 \epsilon_0}}$
 (2) they originate from charges moving with uniform speed.
 (3) they are transverse in nature.
 (4) their energy density in electric field is equal to energy density in magnetic field.

Ans. [2]

Q.39 Choose the correct circuit which can achieve the bridge balance.





Ans. [3]

- Q.40** If the plates of a parallel plate capacitor connected to a battery are moved close to each other, then
- the charge stored in it, increase.
 - the energy stored in it, decreases.
 - its capacitance increases.
 - the ratio of charge to its potential remains the same.
 - the product of charge and voltage increases.

Choose the most appropriate answer from the options given below:

- | | |
|---------------------|---------------------|
| (1) B, D and E only | (2) A, B and C only |
| (3) A, B and E only | (4) A, C and E only |

Ans. [4]

- Q.41** A force defined by $F = \alpha t^2 + \beta t$ acts on a particle at a given time t . The factor which is dimensionless, if α and β are constants, is:

- | | | | |
|---------------------|-----------------------------|------------------------------|------------------------------|
| (1) $\alpha\beta t$ | (2) $\frac{\alpha\beta}{t}$ | (3) $\frac{\beta t}{\alpha}$ | (4) $\frac{\alpha t}{\beta}$ |
|---------------------|-----------------------------|------------------------------|------------------------------|

Ans. [4]

- Q.42** A metallic bar of Young's modulus, $0.5 \times 10^{11} \text{ N m}^{-2}$ and coefficient of linear thermal expansion $10^{-5} \text{ }^\circ\text{C}^{-1}$ length 1 m and area of cross-section 10^{-3} m^2 is heated from 0°C to 100°C without expansion or bending. The compressive force developed in it is:

- | | |
|---------------------------------|--------------------------------|
| (1) $100 \times 10^3 \text{ N}$ | (2) $2 \times 10^3 \text{ N}$ |
| (3) $5 \times 10^3 \text{ N}$ | (4) $50 \times 10^3 \text{ N}$ |

Ans. [4]

- Q.43** A small telescope has an objective of focal length 140 cm and an eye piece of focal length 5.0 cm. The magnifying power of telescope for viewing a distant object is:

- | | | | |
|--------|--------|--------|--------|
| (1) 17 | (2) 32 | (3) 34 | (4) 28 |
|--------|--------|--------|--------|

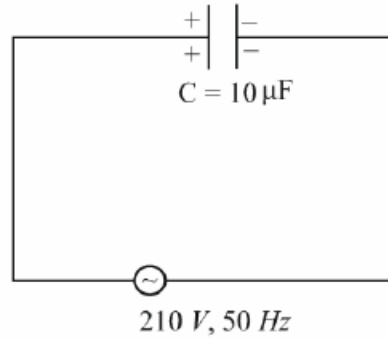
Ans. [4]

- Q.44** An iron bar of length L has magnetic moment M . It is bent at the middle of its length such that the two arms make an angle 60° with each other. The magnetic moment of this new magnet is:

- | | | | |
|----------|--------------------------|---------|-------------------|
| (1) $2M$ | (2) $\frac{M}{\sqrt{3}}$ | (3) M | (4) $\frac{M}{2}$ |
|----------|--------------------------|---------|-------------------|

Ans. [4]

Q.45 A $10 \mu\text{F}$ capacitor is connected to a 210 V , 50 Hz source as shown in figure. The peak current in the circuit is nearly ($\pi = 3.14$):



- (1) 1.20 A (2) 0.35 A (3) 0.58 A (4) 0.93 A

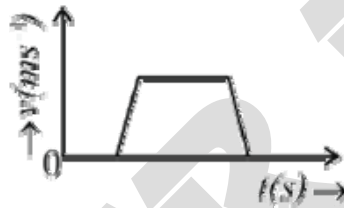
Ans. [4]

Q.46 Two heaters A and B have power rating of 1 kW and 2 kW , respectively. Those two are first connected in series and then in parallel to a fixed power source. The ratio of power outputs for these two cases is:

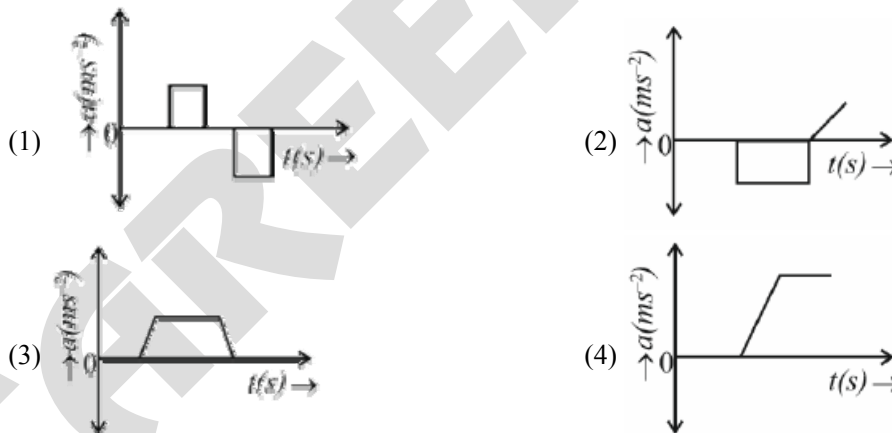
- (1) $1 : 2$ (2) $2 : 3$ (3) $1 : 1$ (4) $2 : 9$

Ans. [4]

Q.47 The velocity (v) – time (t) plot of the motion of a body is shown below:



The acceleration (a) – time (t) graph that best suits this motion is:



Ans. [1]

Q.48 If the mass of the bob in a simple pendulum is increased to thrice its original mass and its length is made half its original length, then the new time period of oscillation is $\frac{x}{2}$ times its original time period. Then the value of x is:

- (1) $2\sqrt{3}$ (2) 4 (3) $\sqrt{3}$ (4) $\sqrt{2}$

Ans. [4]

Q.49 The minimum energy required to launch a satellite of mass m from the surface of earth of mass M and radius R in a circular orbit at an altitude of $2R$ from the surface of the earth is:

- (1) $\frac{GmM}{2R}$ (2) $\frac{GmM}{3R}$ (3) $\frac{5GmM}{6R}$ (4) $\frac{2GmM}{3R}$

Ans. [3]

Q.50 A sheet is placed on a horizontal surface in front of a strong magnetic pole. A force is needed to:

- A. hold the sheet there if it is magnetic.
 B. hold the sheet there if it non-magnetic.
 C. move the sheet away from the pole with uniform velocity if it is conducting.
 D. move the sheet away from the pole with uniform velocity if it is both, non-conducting and non-polar.
 Choose the correct statement(s) from the options given below:

- (1) A, C and D only (2) C only (3) B and D only (4) A and C only

Ans. [4]

CHEMISTRY

SECTION-A

Q.51 Match List with List-II.

	List-I (Conversion)		List-II (Number of Faraday required)
A.	1 mol of H_2O to O_2	I.	3F
B.	1 mol of MnO_4^- to Mn^{2+}	II.	2F
C.	1.5 mol of Ca from molten $CaCl_2$	III.	1F
D.	1 mol of FeO to Fe_2O_3	IV.	5F

Choose the correct answer from the option given below :

- (1) A-II, B-III, C-I, D-IV (2) A-III, B-IV, C-II, D-I
 (3) A-II, B-IV, C-I, D-III (4) A-III, B-IV, C-I, D-II

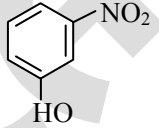
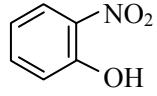
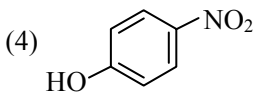
Ans. [3]

Q.52 Which reaction **NOT** a redox reaction ?

- (1) $H_2 + Cl_2 \rightarrow 2HCl$ (2) $BaCl_2 + Na_2SO_4 \rightarrow BaSO_4 + 2 NaCl$
 (3) $Zn + CuSO_4 \rightarrow ZnSO_4 + Cu$ (4) $2 KClO_3 + I_2 \rightarrow 2 KIO_3 + Cl_2$

Ans. [2]

Q.53 Intramolecular hydrogen bonding is present in

- (1)  (2) HF (3)  (4) 

Ans. [3]

Q.54 Fehling's solution 'A' is

- (1) alkaline solution of sodium potassium tartrate (Rochelle's salt)
 (2) aqueous sodium citrate
 (3) aqueous copper sulphate
 (4) alkaline copper sulphate

Ans. [3]

- Q.55** 1 gram of sodium hydroxide was treated with 25 mL of 0.75 M HCl solution, the mass of sodium hydroxide left unreacted is equal to
(1) Zero mg (2) 200 mg (3) 750 mg (4) 250 mg

Ans. [4]

- Q.56** Match the List I with List II.

	List-I (Compound)		List-II (Shape/geometry)
A.	NH ₃	I.	Trigonal Pyramidal
B.	BrF ₅	II.	Square Planar
C.	XeF ₄	III.	Octahedral
D.	SF ₆	IV.	Square Pyramidal

Choose the correct answer from the option given below :

- (1) A-III, B-IV, C-I, D-II (2) A-II, B-III, C-IV, D-I
(3) A-II, B-IV, C-II, D-III (4) A-II, B-IV, C-III, D-I

Ans. [3]

- Q.57** The E° value for the Mn³⁺/Mn²⁺ couple is more positive than that of Cr³⁺/Cr²⁺ or Fe³⁺/Fe²⁺ due to change of
(1) d⁴ to d⁵ configuration (2) d³ to d⁵ configuration
(3) d⁵ to d⁴ configuration (4) d⁵ to d² configuration

Ans. [1]

- Q.58** Match the List I with List II.

	List-I (Process)		List-II (Conditions)
A.	Isothermal process	I.	No heat exchange
B.	Isochoric process	II.	Carried out at constant temperature
C.	Isobaric process	III.	Carried out at constant volume
D.	Adiabatic process	IV.	Carried out at constant pressure

Choose the correct answer from the option given below :

- (1) A-I, B-II, C-III, D-IV (2) A-II, B-III, C-IV, D-I
(3) A-IV, B-III, C-II, D-I (4) A-IV, B-II, C-III, D-I

Ans. [2]

- Q.59** Activation energy of any chemical reaction can be calculated if one knows the value of
(1) orientation of reactant molecules during collision.
(2) rate constant at two different temperatures.
(3) rate constant at standard temperature.
(4) probability of collision.

Ans. [2]

- Q.60** A compound with a molecular formula of C₆H₁₄ has two tertiary carbons. Its IUPAC name is :
(1) 2,3-dimethylbutane (2) 2,2-dimethylbutane
(3) n-hexane (4) 2-methylpentane

Ans. [1]

- Q.69** The reagents with which glucose does **not** react to give the corresponding tests/products are
 A. Tollen's reagent B. Schiff's reagent C. HCN D. NH₂OH
 E. NaHSO₃

Choose the correct options from the given below :

- (1) B and E (2) E and D (3) B and C (4) A and D

Ans. [1]

- Q.70** Match List with List-II.

	List-I (Molecule)		List-II (Number and types of bonds between two carbon atoms)
A.	ethane	I.	one σ -bond and two π -bonds
B.	ethene	II.	two π -bonds
C.	carbon molecule, C ₂	III.	one σ -bond
D.	ethyne	IV.	one σ -bond and one π -bond

Choose the correct answer from the option given below :

- (1) A-III, B-IV, C-II, D-I (2) A-III, B-IV, C-I, D-II
 (3) A-I, B-IV, C-II, D-III (4) A-IV, B-III, C-II, D-I

Ans. [1]

- Q.71** Among Group 16 elements, which one does **NOT** show -2 oxidation state ?

- (1) Te (2) Po (3) O (4) Se

Ans. [2]

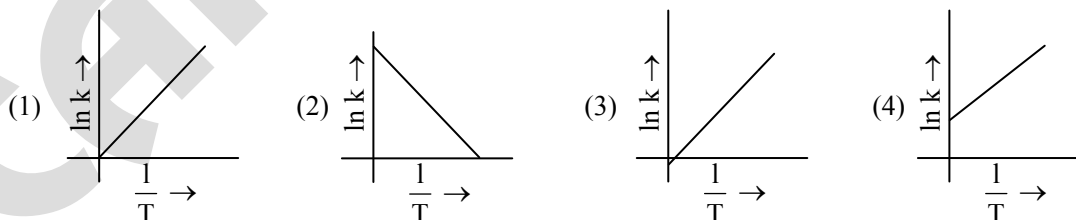
- Q.72** For the reaction $2A \rightleftharpoons B + C$, $K_c = 4 \times 10^{-3}$. At a given time, the composition of reaction mixture is :
 $[A] = [B] = [C] = 2 \times 10^{-3}$ M.

Then which of the following is correct?

- (1) Reaction has a tendency to go in backward direction
 (2) Reaction has gone to completion in forward direction
 (3) Reaction is at equilibrium.
 (4) Reaction has a tendency to go in forward direction.

Ans. [1]

- Q.73** Which plot of $\ln k$ vs $\frac{1}{T}$ is consistent with Arrhenius equation ?



Ans. [2]

- Q.74** In which of the following equilibria, K_p and K_c are **NOT** equal ?

- (1) $CO_{(g)} + H_2O_{(g)} \rightleftharpoons CO_{2(g)} + H_{2(g)}$ (2) $2BrCl_{(g)} \rightleftharpoons Br_{2(g)} + Cl_{2(g)}$
 (3) $PCl_{5(g)} \rightleftharpoons PCl_{3(g)} + Cl_{2(g)}$ (4) $H_{2(g)} + I_{2(g)} \rightleftharpoons 2HI_{(g)}$

Ans. [3]

Q.75 Given below are two statements:

Statement-I : The boiling point of three isomeric pentanes follows the order n-pentane > isopentane > neopentane

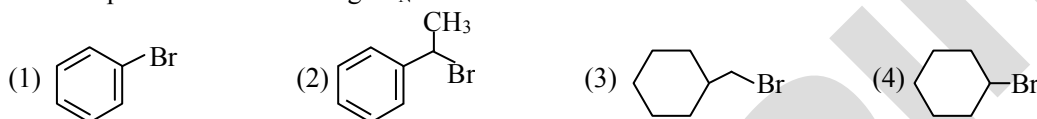
Statement-II : When branching increases, the molecule attains a shape of sphere. This results in smaller surface area for contact, due to which the intermolecular forces between the spherical molecules are weak, thereby lowering the boiling point.

In the light of the above statement, choose the *most appropriate* answer from the options given below:

- (1) Statement I is correct but Statement II is incorrect.
- (2) Statement I is incorrect but Statement II is correct.
- (3) Both Statement I and Statement II are correct.
- (4) Both Statement I and Statement II are incorrect.

Ans. [3]

Q.76 The compound that will undergo S_N^1 reaction with the fastest rate is



Ans. [2]

Q.77 The energy of an electron in the ground state ($n = 1$) for He^+ ion is $-x$ J, then that for an electron in $n = 2$ state for Be^{3+} ion in J is :

- (1) $-4x$
- (2) $-\frac{4}{9}x$
- (3) $-x$
- (4) $-\frac{x}{9}$

Ans. [3]

Q.78 In which of the following processes entropy increases ?

- A. A liquid evaporates to vapour.
- B. Temperature of a crystalline solid lowered from 130 K to 0 K.
- C. $2NaHCO_{3(s)} \rightarrow Na_2CO_{3(s)} + CO_{2(g)} + H_2O_{(g)}$
- D. $Cl_{2(g)} \rightarrow 2 Cl_{(g)}$

Choose the correct answer from the options given below :

- (1) A, C and D
- (2) C and D
- (3) A and C
- (4) A, B and D

Ans. [1]

Q.79 On heating, some solid substances change from solid to vapour state without passing through liquid state. The technique used for the purification of such solid substances based on the above principle is known as

- (1) Distillation
- (2) Chromatography
- (3) Crystallization
- (4) Sublimation

Ans. [4]

Q.80 Match List with List-II.

	List-I (Complex)		List-II (Type of isomerism)
A.	$[Co(NH_3)_5(NO_2)]Cl_2$	I.	Solvate isomerism
B.	$[Co(NH_3)_5(SO_4)]Br$	II.	Linkage isomerism
C.	$[Co(NH_3)_6][Cr(CN)_6]$	III.	Ionization isomerism
D.	$[Co(H_2O)_6]Cl_3$	IV.	Coordination isomerism

Choose the correct answer from the option given below :

- (1) A-I, B-IV, C-III, D-II
- (2) A-II, B-IV, C-III, D-I
- (3) A-II, B-III, C-IV, D-I
- (4) A-I, B-III, C-IV, D-II

Ans. [3]

Q.81 Given below are two statements :

Statement-I : Aniline does not undergo Friedel-Crafts alkylation reaction.

Statement-II : Aniline cannot be prepared through Gabriel synthesis.

In the light of the above statement, choose the *correct* answer from the options given below:

- (1) Statement I is correct but Statement II is false. (2) Statement I is incorrect but Statement II is true.
(3) Both Statement I and Statement II are true. (4) Both Statement I and Statement II are false.

Ans. [3]

Q.82 Arrange the following elements in increasing order of first ionization enthalpy :

Li, Be, B, C, N

Choose the correct answer from options given below:

- (1) $Li < Be < C < B < N$ (2) $Li < Be < N < B < C$
(3) $Li < Be < B < C < N$ (4) $Li < B < Be < C < N$

Ans. [4]

Q.83 The highest number of helium atoms is in

- (1) 4 g of helium (2) 2.271098 L of helium at STP
(3) 4 mol of helium (4) 4 u of helium

Ans. [3]

Q.84 The most stable carbocation among the following is :



Ans. [2]

Q.85 The Henry's law constant (K_H) values of three gases (A, B, C) in water are 145, 2×10^{-5} and 35 kbar, respectively. The solubility of these gases in water follow the order :

- (1) $A > C > B$ (2) $A > B > C$ (3) $B > A > C$ (4) $B > C > A$

Ans. [4]

SECTION-B

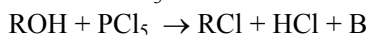
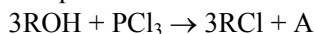
Q.86 A compound X contains 32% of A, 20% of B and remaining percentage of C. Then, the empirical formula of X is :

(Given atomic masses of A = 64; B = 40; C = 32 u)

- (1) AB_2C_2 (2) ABC_4 (3) A_2BC_2 (4) ABC_3

Ans. [4]

Q.87 The products A and B obtained in the following reactions, respectively, are



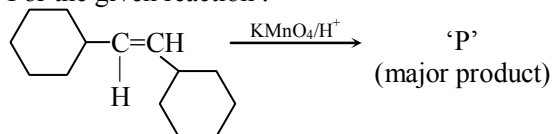
- (1) H_3PO_4 and $POCl_3$ (2) H_3PO_3 and $POCl_3$ (3) $POCl_3$ and H_3PO_4 (4) $POCl_3$ and H_3PO_4

Ans. [2]

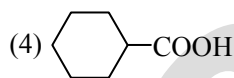
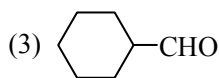
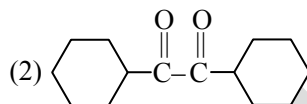
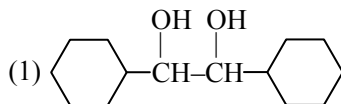
- Q.88** The plot of osmotic pressure (Π) vs concentration (mol L^{-1}) for a solution gives a straight line with slope $25.73 \text{ L bar mol}^{-1}$. The temperature at which the osmotic pressure measurement is done is :
(Use $R = 0.083 \text{ L bar mol}^{-1} \text{ K}^{-1}$)
(1) 25.73°C (2) 12.05°C (3) 37°C (4) 310°C

Ans. [3]

- Q.89** For the given reaction :



'P' is



Ans. [4]

- Q.90** Given below are two statement :

Statement-I : $[\text{Co}(\text{NH}_3)_6]^{3+}$ is a homoleptic complex whereas $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]^+$ is a heteroleptic complex.

Statement-II : Complex $[\text{Co}(\text{NH}_3)_6]^{3+}$ has only one kind of ligands but $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]^+$ has more than one kind of ligands.

In the light of the above statements, choose the *correct* answer from the options given below :

- (1) Statement I true but Statement II is false. (2) Statement I is false but Statement II is true.
(3) Both Statement I and Statement II are true. (4) Both Statement I and Statement II are false.

Ans. [3]

- Q.91** During the preparation of Mohr's salt solution (Ferrous ammonium sulphate), which of the following acid is added to prevent hydrolysis of Fe^{2+} ion ?

- (1) dilute nitric acid (2) dilute sulphuric acid
(3) dilute hydrochloric acid (4) concentrated sulphuric acid

Ans. [2]

- Q.92** Identify the **correct** answer.

- (1) Dipole moment of NF_3 is greater than that of NH_3 .
(2) Three canonical forms can be drawn for CO_3^{2-} ion.
(3) Three resonance structures can be drawn for ozone.
(4) BF_3 has non-zero dipole moment.

Ans. [2]

- Q.93** Given below are certain cations. Using inorganic qualitative analysis, arrange them in increasing group number from 0 to VI.

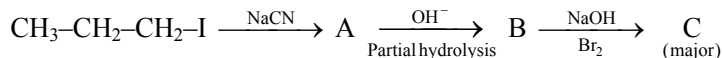
- A. Al^{3+} B. Cu^{2+} C. Ba^{2+} D. Co^{2+}
E. Mg^{2+}

Choose the correct answer from the options given below :

- (1) E, C, D, B, A (2) E, A, B, C, D (3) B, A, D, C, E (4) B, C, A, D, E

Ans. [3]

Q.94 Identify the major product C formed in the following reaction sequence :



- (1) butanamide (2) α -bromobutanoic acid
 (3) propylamine (4) butylamine

Ans. [3]

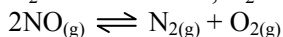
Q.95 The rate of a reaction quadruples when temperature changes from 27°C to 57°C. Calculate the energy of activation.

Given $R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$, $\log 4 = 0.6021$

- (1) 3.80 kJ/mol (2) 3804 kJ/mol (3) 38.04 kJ/mol (4) 380.4 kJ/mol

Ans. [3]

Q.96 Consider the following reaction in a sealed vessel at equilibrium with concentrations of $\text{N}_2 = 3.0 \times 10^{-3} \text{ M}$, $\text{O}_2 = 4.2 \times 10^{-3} \text{ M}$ and $\text{NO} = 2.8 \times 10^{-3} \text{ M}$.



If 0.1 mol L^{-1} of $\text{NO}_{(g)}$ is taken in closed vessel, what will be degree of dissociation (α) of $\text{NO}_{(g)}$ at equilibrium ?

- (1) 0.8889 (2) 0.717 (3) 0.00889 (4) 0.0889

Ans. [2]

Q.97 The work done during reversible isothermal expansion of one mole of hydrogen gas at 25°C from pressure of 20 atmosphere to 10 atmosphere is :

(Given $R = 2.0 \text{ cal K}^{-1} \text{ mol}^{-1}$)

- (1) 413.14 calories (2) 100 calories (3) 0 calories (4) -413.14 calories

Ans. [4]

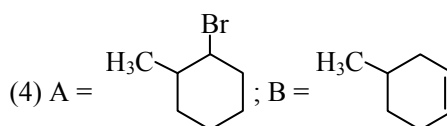
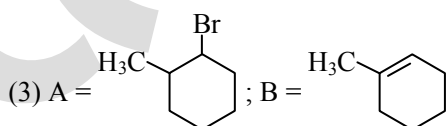
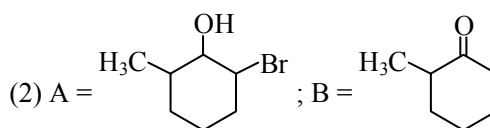
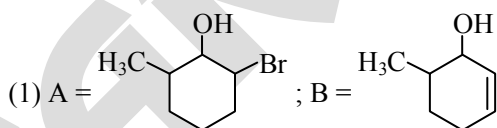
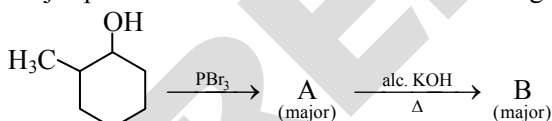
Q.98 Mass in grams of copper deposited by passing 9.6487 A current through a voltmeter containing copper sulphate solution for 100 seconds is :

(Given : Molar mass of Cu : 63 g mol^{-1} , $1F = 96487 \text{ C}$)

- (1) 31.5 g (2) 0.0315 g (3) 3.15 g (4) 0.315 g

Ans. [4]

Q.99 Major product A and B formed in the following reaction sequence, are



Ans. [3]

Q.100 The pair of lanthanoid ions which are diamagnetic is :

- (1) Gd^{3+} and Eu^{3+} (2) Pm^{3+} and Sm^{3+} (3) Ce^{4+} and Yb^{2+} (4) Ce^{3+} and Eu^{2+}

Ans. [3]

BIOLOGY**BOTANY | SECTION-A**

Q.101 Identify the set of correct statements :

- (A) The flowers of Vallisneria are colourful and produce nectar.
- (B) The flower of waterlily are not pollinated by water.
- (C) In most of water-pollinated species, the pollen grains are protected from wetting.
- (D) Pollen grains of some hydrophytes are long and ribbon like.
- (E) In some hydrophytes, the pollen grains are carried passively inside water

Choose the correct answer from the options given below :

- (1) A, C, D and E only (2) B, C, D and E only (3) C, D and E only (4) A, B, C and D only

Ans. [2]

Q.102 The type of conservation in which the treated species are taken out from their naturally habitat and placed in special setting where they can be protected and given special care is called

- (1) Semi-conservative method
- (2) Sustainable development
- (3) in-situ conservation
- (4) Biodiversity conservation

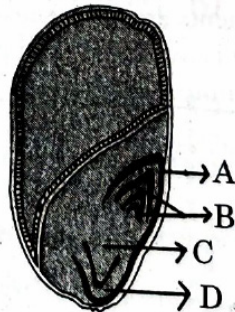
Ans. [4]

Q.103 Inhibition of Succinic dehydrogenase enzyme by malonate is a classical example of

- (1) Competitive inhibition
- (2) Enzyme activation
- (3) Cofactor inhibition
- (4) Feedback inhibition

Ans. [1]

Q.104 Identify the part of the seed from the given figure which is destined to form root when the seed germinates.



- (1) C (2) D (3) A (4) B

Ans. [1]

Q.105 Buliform cells are responsible for

- (1) Increased photosynthesis in monocots
- (2) Providing large spaces for storage of sugar
- (3) Inward curling of leaves in monocots
- (4) Protecting the plant from salt stress

Ans. [3]

Q.106 Which of the following are required for the dark reaction of photosynthesis?

- (A) Light
- (B) Chlorophyll
- (C) CO₂
- (D) ATP
- (E) NADPH

Choose the correct answer from the options given below :

- (1) C, D and E only (2) D and E only (3) A, B and C only (4) B, C and D only

Ans. [1]

Q.107 Formation of interfascicular cambium from fully developed parenchyma cells is an example for
(1) Dedifferentiation (2) Maturation (3) Differentiation (4) Redifferentiation

Ans. [1]

Q.108 Hind II always cuts DNA molecules at a particular point called recognition sequence and it consists of :
(1) 4 bp (2) 10 bp (3) 8 bp (4) 6 bp

Ans. [4]

Q.109 Tropical regions show greatest level of species richness because
(A) Tropical latitudes have remained relatively undisturbed for millions of years, hence more time was available for species diversification.
(B) Tropical environments are more seasonal
(C) More solar energy is available in tropics
(D) Constant environments promote niche specialization.
(E) Tropical environments are constant and predictable.

Choose the correct answer form the options given below :

(1) A, B and E only (2) A, B and D only (3) A, C, D and E only (4) A and B only

Ans. [3]

Q.110 Which one of the following is not a criterion for classification of fungi?

(1) Mode of spore formation (2) Fruiting body
(3) Morphology of mycelium (4) Mode of nutrition

Ans. [4]

Q.111 How many molecules of ATP and NADPH are required for every molecule of CO₂ fixed in the calvin cycle?

(1) 3 molecules of ATP and 3 molecules of NADPH
(2) 3 molecules of ATP and 2 molecules of NADPH
(3) 2 molecules of ATP and 3 molecules of NADPH
(4) 2 molecules of ATP and 2 molecules of NADPH

Ans. [2]

Q.112 These are regarded as major causes of biodiversity loss :

(A) Over exploitation (B) Co-extinction (C) Mutation
(D) Habitat loss and fragmentation (E) Migration

Choose the correct option :

(1) A, B and E only (2) A, B and D only (3) A, C and D only (4) A, B, C and D only

Ans. [2]

Q.113 The capacity to generate a whole plant from any cell of the plant is called :

(1) Differentiation (2) Somatic hybridization
(3) Totipotency (4) Micropropagation

Ans. [3]

Q.114 The equation of Verhulst-Pearl logistic growth is $\frac{dN}{dt} = rN \left[\frac{K - N}{K} \right]$.

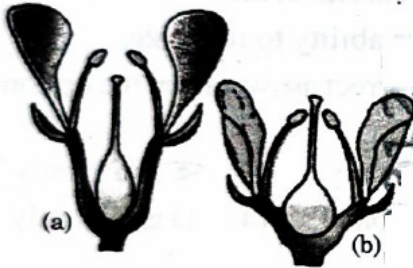
From this equation, K indicates :

(1) Carrying capacity (2) Population density
(3) Intrinsic rate of natural increase (4) Biotic potential

Ans. [1]

- Q.115** Spindle fibers attach to kinetochores of chromosomes during
 (1) Anaphase (2) Telophase (3) Prophase (4) Metaphase
Ans. [4]

- Q.116** Identify the type of flowers based on the position of calyx, corolla and androecium with respect to the ovary from the given figure (a) and (b)



- (1) (a) Perigynous; (b) Epigynous
 (2) (a) Perigynous; (b) Perigynous
 (3) (a) Epigynous; (b) Hypogynous
 (4) (a) Hypogynous; (b) Epigynous
Ans. [2]

- Q.117** Match List I with List II

List I

- A. Rhizopus
 B. Ustilago
 C. Puccinia
 D. Agaricu

List II

- I. Mushroom
 II. Smut fungus
 III. Bread mould
 IV. Rust fungus

Choose the correct answer from the options given below :

- (1) A-III, B-II, C-I, D-IV
 (2) A-IV, B-III, C-II, D-I
 (3) A-III, B-II, C-IV, D-I
 (4) A-I, B-III, C-II, D-IV
Ans. [3]

- Q.118** In a plant, black seed color (BB/Bb) is dominant over white seed color (bb). In order to find out the genotype of the black seed plant, with which of the following genotype will you cross it?

- (1) Bb (2) BB/Bb (3) BB (4) bb
Ans. [4]

- Q.119** A pink flowered Snapdragon plant was crossed with a red flowered snapdragon plant. What type of phenotype/s is/are expected in the progeny?

- (1) Only pink flowered plants
 (2) Red, pink as well as white flowered plants
 (3) Only red flowered plants
 (4) Red flowered as well as pink flowered plants
Ans. [4]

- Q.120** Match List I with List II

List I	List II
(A) Two or more alternative forms of a gene	(I) Back cross
(B) Cross of F ₁ progeny with homozygous recessive parent	(II) Ploidy
(C) Cross of F ₁ progeny with any of the parents	(III) Allele
(D) Number of chromosome sets in plant	(IV) Test cross

Choose the correct answer from the options given below :

- (1) A-III, B-IV, C-I, D-II
 (2) A-IV, B-III, C-II, D-I
 (3) A-I, B-II, C-III, D-IV
 (4) A-II, B-I, C-III, D-IV
Ans. [1]

Q.121 Lecithin, a small molecular weight organic compound found in living tissues, is an example of :
 (1) Glycerides (2) Carbohydrates (3) Amino acids (4) Phospholipids

Ans. [4]

Q.122 Match List I with List II

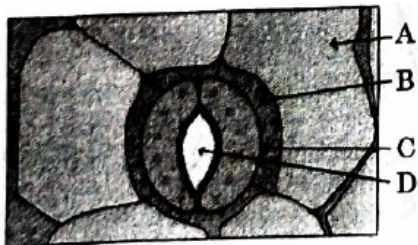
List I	List II
(A) Clostridium butylicum	(I) Ethanol
(B) Saccharomyces cerevisiae	(II) Streptokinase
(C) Trichoderma polysporum	(III) Butyric acid
(D) Streptococcus sp	(IV) Cyclosporin-A

Choose the correct answer from the options given below :

- (1) A-III, B-I, C-IV, D-II (2) A-IV, B-I, C-III, D-II
 (3) A-III, B-I, C-II, D-IV (4) A-II, B-IV, C-III, D-I

Ans. [1]

Q.123 In the given figure, which component has thin outer walls and highly thickened inner walls?



- (1) A (2) B (3) C (4) D

Ans. [3]

Q.124 Which of the following is an example of actinomorphic flower?

- (1) Pisum (2) Sesbania (3) Datura (4) Cassia

Ans. [3]

Q.125 A transcription unit in DNA is defined primarily by the three regions in DNA and these are with respect to upstream and down stream end;

- (1) Inducer, Repressor, Structural gene (2) Promotor, Structural gene, Terminator
 (3) Repressor, Operator gene, Structural gene (4) Structural gene, Transposons, Operator gene

Ans. [2]

Q.126 What is the fate of piece of DNA carrying only gene of interest which is transferred into an alien organism?

- A. The piece of DNA would be able to multiply itself independently in the progeny cells of the organisms.
 B. It may get generated into the genome of the recipient.
 C. It may multiply and be inherited along with the host DNA.
 D. The alien piece of DNA is not an integral part of chromosome.
 E. It shows ability to replicate.

Choose the correct answer from the options given below:

- (1) B and C only (2) A and E only
 (3) A and B only (4) D and E only

Ans. [1]

- Q.127** Auxin is used by gardeners to prepare weed free lawns. But no damage is caused to grass as auxin;
- (1) does not affect mature monocotyledonous plants.
 - (2) can help in cell division in grasses, to produce growth.
 - (3) promotes apical dominance.
 - (4) promotes abscission of mature leaves only.

Ans. [1]

- Q.128** The cofactor of the enzyme carboxypeptidase is
- (1) Flavin (2) Haem (3) Zinc (4) Niacin

Ans. [3]

- Q.129** The lactose present in the growth medium of bacteria is transported to the cell by the action of
- (4) Permease (2) Polymerase (3) Beta-galactosidase (4) Acetylase

Ans. [1]

- Q.130** Which one of the following can be explained on the basis of Mendel's Law of Dominance?
- (A) Out of one pair of factors one is dominant and the other is recessive.
(B) Alleles do not show any expression and both the characters appear as such in F₂ generation
(C) Factors occur in pairs in normal diploid plants
(D) The discrete unit controlling a particular character is called factor
(E) The expression of only one of the parental characters is found in a monohybrid cross.

Choose the correct answer from the options given below :

- (1) B, C and D only (2) A, B, C, D and E (3) A, B and C only (4) A, C, D and E only

Ans. [4]

- Q.131** Given below are two statements :

Statement I : Bt toxins are insect group specific and coded by a gene cry IAc.

Statement II : Bt toxin exists as inactive protoxin in *B. thuringiensis*. However, after ingestion by the insect the inactive protoxin gets converted into active form due to acidic pH of the insect gut.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false
- (2) Statement I is false but Statement II is true
- (3) Both Statement I and Statement II are true
- (4) Both Statement I and Statement II are false

Ans. [1]

- Q.132** Given below are two statements :

Statement I : Parenchyma is living but collenchyma is dead tissue

Statement II: Gymnosperms lack xylem vessels but presence of xylem vessels is the characteristic of angiosperms.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false.
- (2) Statement I is false but Statement II is true.
- (3) Both Statement I and Statement II are true.
- (4) Both Statement I and Statement II are false

Ans. [2]

Q.133 Given below are two Statements:

Statement I : Chromosomes become gradually visible under light microscope during leptotene stage.

Statement II: The beginning of diplotene stage is recognized by dissolution of synaptonemal complex.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false.
- (2) Statement I is false but Statement II is true.
- (3) Both Statement I and Statement II are true.
- (4) Both Statement I and Statement II are false

Ans. [3]

Q.134 Match List-I with List-II.

List I	List II
(A) Nucleolus	(I) Site of formation of glycolipid
(B) Centriole	(II) Organization like the cartwheel
(C) Leucoplasts	(III) Site for active ribosomal RNA synthesis
(D) Golgi apparatus	(IV) For storing nutrients

Choose the correct answer from the options given below

- (1) A-III, B-IV, C-II, D-I
- (2) A-I, B-II, C-III, D-IV
- (3) A-III, B-II, C-IV, D-I
- (4) A-II, B-III, C-I, D-IV

Ans. [3]

Q.135 List of endangered species was released by:-

- (1) Foam
- (2) IUCN
- (3) GEAC
- (4) WWF

Ans. [2]

BOTANY | SECTION-B

Q.136 The DNA present in chloroplast is:

- (1) Linear, single stranded
- (2) Circular, single stranded
- (3) Linear, double stranded
- (4) Circular, double stranded

Ans. [4]

Q.137 Which of the following are fused in somatic hybridization involving two varieties of plants?

- (1) Protoplasts
- (2) Pollens
- (3) Callus
- (4) Somatic embryos

Ans. [1]

Q.138 Identify the correct description about the given figure:



- (1) Cleistogamous flowers showing autogamy.
- (2) Compact inflorescence showing complete autogamy.
- (3) Wind pollinated plant inflorescence showing flowers with well exposed stamens.
- (4) Water pollinated flowers showing stamens with mucilaginous covering.

Ans. [3]

Q.139 Spraying sugarcane crop with which of the following plant growth regulators, increases the length of stem, thus, increasing the yield?

- (1) Cytokinin (2) Abscisic acid (3) Auxin (4) Gibberellin

Ans. [4]

Q.140 Match List-I with List-II.

List I	List II
(A) Frederick Griffith	(I) Genetic code
(B) Francois Jacob & Jacque Monod	(II) Semi-conservative mode of DNA replication
(C) Har Gobind Khorana	(III) Transformation
(D) Meselson & Stahl	(IV) Lac operon

Choose the correct answer from the options given below

- (1) A-II, B-III, C-IV, D-I (2) A-IV, B-I, C-II, D-III
(3) A-III, B-II, C-I, D-IV (4) A-III, B-IV, C-I, D-II

Ans. [4]

Q.141 Match List-I with List-II.

List I	List II
(A) GLUT-4	(I) Hormone
(B) Insulin	(II) Enzyme
(C) Trypsin	(III) Intercellular ground substances
(D) Collagen	(IV) Enables glucose transport into cell.

Choose the correct answer from the options given below

- (1) A-II, B-III, C-IV, D-I (2) A-III, B-IV, C-I, D-II
(3) A-IV, B-I, C-II, D-III (4) A-I, B-II, C-III, D-IV

Ans. [3]

Q.142 Given below are two statements:

Statement I : In C_3 Plants, some O_2 binds RuBisCO, hence CO_2 fixation is decreased.

Statement II: In C_4 plants, mesophyll cells show very little photorespiration while bundle sheath cells do not show photorespiration.

In the light of the above statements, choose the correct answer from the options given below:

- (1) Statement I is true but Statement II is false.
(2) Statement I is false but Statement II is true.
(3) Both Statement I and Statement II are true.
(4) Both Statement I and Statement II are false

Ans. [1]

Q.143 Identify the step in tricarboxylic acid cycle, which does not involve oxidation of substrate.

- (1) Succinyl-CoA \rightarrow Succinic acid
(2) Isocitrate \rightarrow α -ketoglutaric acid
(3) Malic acid \rightarrow Oxaloacetic acid
(4) Succinic acid \rightarrow Malic acid

Ans. [1]

Q.153 Given below are two statements :

Statement I: The presence or absence of hymen is not a reliable indicator of virginity.

Statement II : The hymen is torn during the first coitus only.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false (2) Statement I is false but Statement II is true
 (3) Both Statement I and Statement II are true (4) Both Statement I and Statement II are false

Ans. [1]

Q.154 In both sexes of cockroach, a pair of jointed filamentous structures called anal cerci are present on:

- (1) 8th and 9th segment (2) 11th segment (3) 5th segment (4) 10th segment

Ans. [4]

Q.155 Match List I with List II :

	List I		List II
A.	Pons	I.	Provides additional space for Neurons, regulates posture and balance.
B.	Hypothalamus	II.	Controls respiration and gastric secretions.
C.	Medulla	III.	Connects different regions of the brain.
D.	Cerebellum	IV.	Neuro secretory cells

Choose the correct answer from the options given below:

- (1) A-I, B-III, C-II, D-IV (2) A-II, B-I, C-III, D-IV
 (3) A-II, B-III, C-I, D-IV (4) A-III, B-IV, C-II, D-I

Ans. [4]

Q.156 Which of the following is not a steroid hormone?

- (1) Progesterone (2) Glucagon (3) Cortisol (4) Testosterone

Ans. [2]

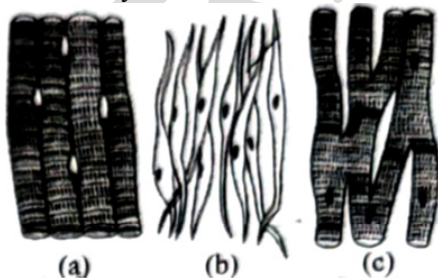
Q.157 Which one is the correct product of DNA dependent RNA polymerase to the given template?

3'TACATGGCAAATATCCATTCA5'

- (1) 5'AUGUACCGUUUAUAGGGAAGU3' (2) 5'ATGTACCGTTTATAGGTAAGT3'
 (3) 5'AUGUACCGUUUAUAGGUAAGU3' (4) 5'AUGUAAAGUUUAUAGGUAAGU3'

Ans. [3]

Q.158 Three types of muscles are given as a, b and c. Identify the correct matching pair along with their location in human body :



Name of muscle/location

- (1) (a) Skeletal - Biceps; (b) Involuntary - Intestine; (c) Smooth - Heart.
 (2) (a) Involuntary - Nose tip; (b) Skeletal - Bone; (c) Cardiac - Heart.
 (3) (a) Smooth - Toes; (b) Skeletal - Legs; (c) Cardiac - Heart.
 (4) (a) Skeletal - Triceps; (b) Smooth - Stomach; (c) Cardiac - Heart.

Ans. [4]

Q.159 Following are the stages of cell division :

- A. Gap 2 phase
- B. Cytokinesis
- C. Synthesis phase
- D. Karyokinesis
- E. Gap 1 phase

Choose the correct sequence of stages from the options given below :

- (1) B-D-E-A-C (2) E-C-A-D-B (3) C-E-D-A-B (4) E-B-D-A-C

Ans. [2]

Q.160 Which of the following are Autoimmune disorders?

- A. Myasthenia gravis
- B. Rheumatoid arthritis
- C. Gout
- D. Muscular dystrophy
- E. Systemic Lupus Erythematosus (SLE)

Choose the most appropriate answer from the options given below :

- (1) B, C & E only (2) C, D & E only (3) A, B & D only (4) A, B & E only

Ans. [4]

Q.161 Match List I with List II :

	List I		List II
A.	Lipase	I.	Peptide bond
B.	Nuclease	II.	Ester bond
C.	Protease	III.	Glycosidic bond
D.	Amylase	IV.	Phosphodiester bond

Choose the correct answer from the options given below:

- (1) A-II, B-IV, C-I, D-III (2) A-IV, B-I, C-III, D-II
(3) A-IV, B-II, C-III, D-I (4) A-III, B-II, C-I, D-IV

Ans. [1]

Q.162 The flippers of the Penguins and Dolphins are the example of the

- (1) Convergent evolution (2) Divergent evolution
(3) Adaptive radiation (4) Natural selection

Ans. [1]

Q.163 Match List I with List II :

	List I		List II
A.	Expiratory capacity	I.	Expiratory reserve volume + Tidal volume + Inspiratory reserve volume
B.	Functional residual capacity	II.	Tidal volume + Expiratory reserve volume
C.	Vital capacity	III.	Tidal volume + Inspiratory reserve volume
D.	Inspiratory capacity	IV.	Expiratory reserve volume + Residual volume

Choose the correct answer from the options given below :

- (1) A-II, B-I, C-IV, D-III (2) A-I, B-III, C-II, D-IV
(3) A-II, B-IV, C-I, D-III (4) A-III, B-II, C-IV, D-I

Ans. [3]

Q.164 Which one of the following factors will not affect the Hardy-Weinberg equilibrium?
(1) Gene migration (2) Constant gene pool (3) Genetic recombination (4) Genetic drift
Ans. [2]

Q.165 Given below are some stages of human evolution. Arrange them in correct sequence. (Past to Recent)
A. *Homo habilis*
B. *Homo sapiens*
C. *Homo neanderthalensis*
D. *Homo erectus*
Choose the correct sequence of human evolution from the options given below :
(1) C-B-D-A (2) A-D-C-B (3) D-A-C-B (4) B-A-D-C
Ans. [2]

Q.166 Following are the stages of pathway for conduction of an action potential through the heart :
A. AV bundle
B. Purkinje fibres
C. AV node
D. Bundle branches
E. SA node
Choose the correct sequence of pathway from the options given below :
(1) B-D-E-C-A (2) E-A-D-B-C (3) E-C-A-D-B (4) A-E-C-B-D
Ans. [3]

Q.167 Which of the following factors are favourable for the formation of oxyhaemoglobin in alveoli ?
(1) Low $p\text{CO}_2$ and High H^+ concentration (2) Low $p\text{CO}_2$ and High temperature
(3) High $p\text{O}_2$ and High $p\text{CO}_2$ (4) High $p\text{O}_2$ and Lesser H^+ concentration
Ans. [4]

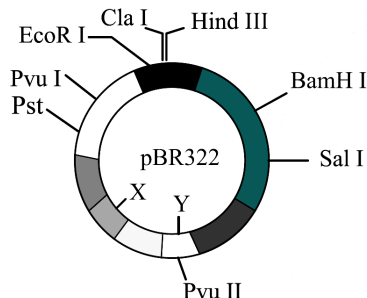
Q.168 Match List I with List II:

	List I		List II
A.	α -1 antitrypsin	I.	Cotton bollworm
B.	Cry IAb	II.	ADA deficiency
C.	Cry IAc	III.	Emphysema
D.	Enzyme replacement therapy	IV.	Corn borer

Choose the correct answer from the options given below :
(1) A-III, B-IV, C-I, D-II (2) A-II, B-IV, C-I, D-III
(3) A-II, B-I, C-IV, D-III (4) A-III, B-I, C-II, D-IV
Ans. [1]

Q.169 Given below are two statements: one is labelled as Assertion A and the other is labelled as Reason R :
Assertion A : FSH acts upon ovarian follicles in female and Leydig cells in male.
Reason R : Growing ovarian follicles secrete estrogen in female while interstitial cells secrete androgen in male human being.
In the light of the above statements, choose the correct answer from the options given below :
(1) A is true but R is false
(2) A is false but R is true
(3) Both A and R are true and R is the correct explanation of A.
(4) Both A and R are true but R is NOT the correct explanation of A.
Ans. [2]

Q.170 The following diagram showing restriction sites in *E. coli* cloning vector pBR322. Find the role of 'X' and 'Y' genes :



- (1) The gene 'X' is for protein involved in replication of Plasmid and 'Y' for resistance to antibiotics.
- (2) Gene 'X' is responsible for recognition sites and 'Y' is responsible for antibiotic resistance.
- (3) The gene 'X' is responsible for resistance to antibiotics and 'Y' for protein involved in the replication of Plasmid.
- (4) The gene 'X' is responsible for controlling the copy number of the linked DNA and 'Y' for protein involved in the replication of Plasmid.

Ans. [4]

Q.171 Match List I with List II:

	List I		List II
A.	Cocaine	I.	Effective sedative in surgery
B.	Heroin	II.	<i>Cannabis sativa</i>
C.	Morphine	III.	<i>Erythroxylum</i>
D.	Marijuana	IV.	<i>Papaver somniferum</i>

Choose the correct answer from the options given below:

- (1) A-II, B-I, C-III, D-IV
- (2) A-III, B-IV, C-I, D-II
- (3) A-IV, B-III, C-I, D-II
- (4) A-I, B-III, C-II, D-IV

Ans. [2]

Q.172 Consider the following statements :

- A. Annelids are true coelomates
- B. Poriferans are pseudocoelomates
- C. Aschelminthes are acoelomates
- D. Platyhelminthes are pseudocoelomates

Choose the correct answer from the options given below :

- (1) C only
- (2) D only
- (3) B only
- (4) A only

Ans. [3]

Q.173 Given below are two statements :

Statement I: In the nephron, the descending limb of loop of Henle is impermeable to water and permeable to electrolytes.

Statement II: The proximal convoluted tubule is lined by simple columnar brush border epithelium and increases the surface area for reabsorption.

In the light of the above statements, choose the correct answer from the options given below :

- (1) Statement I is true but Statement II is false
- (2) Statement I is false but Statement II is true
- (3) Both Statement I and Statement II are true
- (4) Both Statement I and Statement II are false

Ans. [4]

- Q.183** The “Ti plasmid” of *Agrobacterium tumefaciens* stands for
 (1) Tumor inducing plasmid (2) Temperature independent plasmid
 (3) Tumour, inhibiting plasmid (4) Tumor independent plasmid

Ans. [1]

- Q.184** Which of the following is not a component of Fallopian tube?
 (1) Infundibulum (2) Ampulla (3) Uterine fundus (4) Isthmus

Ans. [3]

- Q.185** Match List I with List II:

	List I		List II
A.	Down's syndrome	I.	11 th chromosome
B.	α -Thalassemia	II.	'X' chromosome
C.	β -Thalassemia	III.	21 st chromosome
D.	Klinefelter's syndrome	IV.	16 th chromosome

Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-I, D-II (2) A-IV, B-I, C-II, D-III
 (3) A-I, B-II, C-III, D-IV (4) A-II, B-III, C-IV, D-I

Ans. [1]

ZOOLOGY | SECTION-B

- Q.186** The following are the statements about non-chordates:

- A. Pharynx is perforated by gill slits.
 B. Notochord is absent.
 C. Central nervous system is dorsal.
 D. Heart is dorsal if present.
 E. Post anal tail is absent.

Choose the most appropriate answer from the options given below :

- (1) B, D & E only (2) B, C & D only (3) A & C only (4) A, B & D only

Ans. [1]

- Q.187** Match List I with List II:

	List I		List II
A.	Mesozoic Era	I.	Lower invertebrates
B.	Proterozoic Era	II.	Fish & Amphibia
C.	Cenozoic Era	III.	Birds & Reptiles
D.	Paleozoic Era	IV.	Mammals

Choose the correct answer from the options given below:

- (1) A-I, B-II, C-IV, D-III (2) A-III, B-I, C-IV, D-II
 (3) A-II, B-I, C-III, D-IV (4) A-III, B-I, C-II, D-IV

Ans. [2]

Q.188 Given below are two statements :

Statement I : The cerebral hemispheres are connected by nerve tract known as corpus callosum.

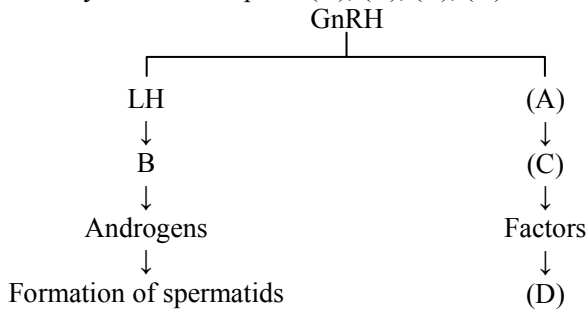
Statement II : The brain stem consists of the medulla oblongata, pons and cerebrum.

In the light of the above statements, choose the most appropriate answer from the options given below:

- (1) Statement I is correct but Statement II is incorrect.
- (2) Statement I is incorrect but Statement II is correct.
- (3) Both Statement I and Statement II are correct.
- (4) Both Statement I and Statement II are incorrect.

Ans. [1]

Q.189 Identify the correct option (A), (B), (C), (D) will respect to spermatogenesis.



- (1) FSH, Sertoli cells, Leydig cells, spermatogenesis.
- (2) ICSH, Leydig cells, Sertoli cells, spermatogenesis
- (3) FSH, Leydig cells, Sertoli cells, spermiogenesis
- (4) ICSH, Interstitial cells, Leydig cells, spermiogenesis

Ans. [3]

Q.190 Match List I with List II :

	List I		List II
A.	RNA polymerase III	I.	snRNPs
B.	Termination of transcription	II.	Promotor
C.	Splicing of Exons	III.	Rho factor
D.	TATA box	IV.	SnRNAs, tRNA

Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-I, D-II
- (2) A-IV, B-III, C-I, D-II
- (3) A-II, B-IV, C-I, D-III
- (4) A-III, B-II, C-IV, D-I

Ans. [2]

Q.191 Match List I with List II:

	List I		List II
A.	Exophthalmic goiter	I.	Excess secretion of cortisol, moon face and hyperglycemia
B.	Acromegaly	II.	Hypo-secretion of thyroid hormone and stunted growth.
C.	Cushing's syndrome	III.	Hyper secretion of thyroid hormone and protruding eye balls.
D.	Cretinism	IV.	Excessive secretion of growth hormone.

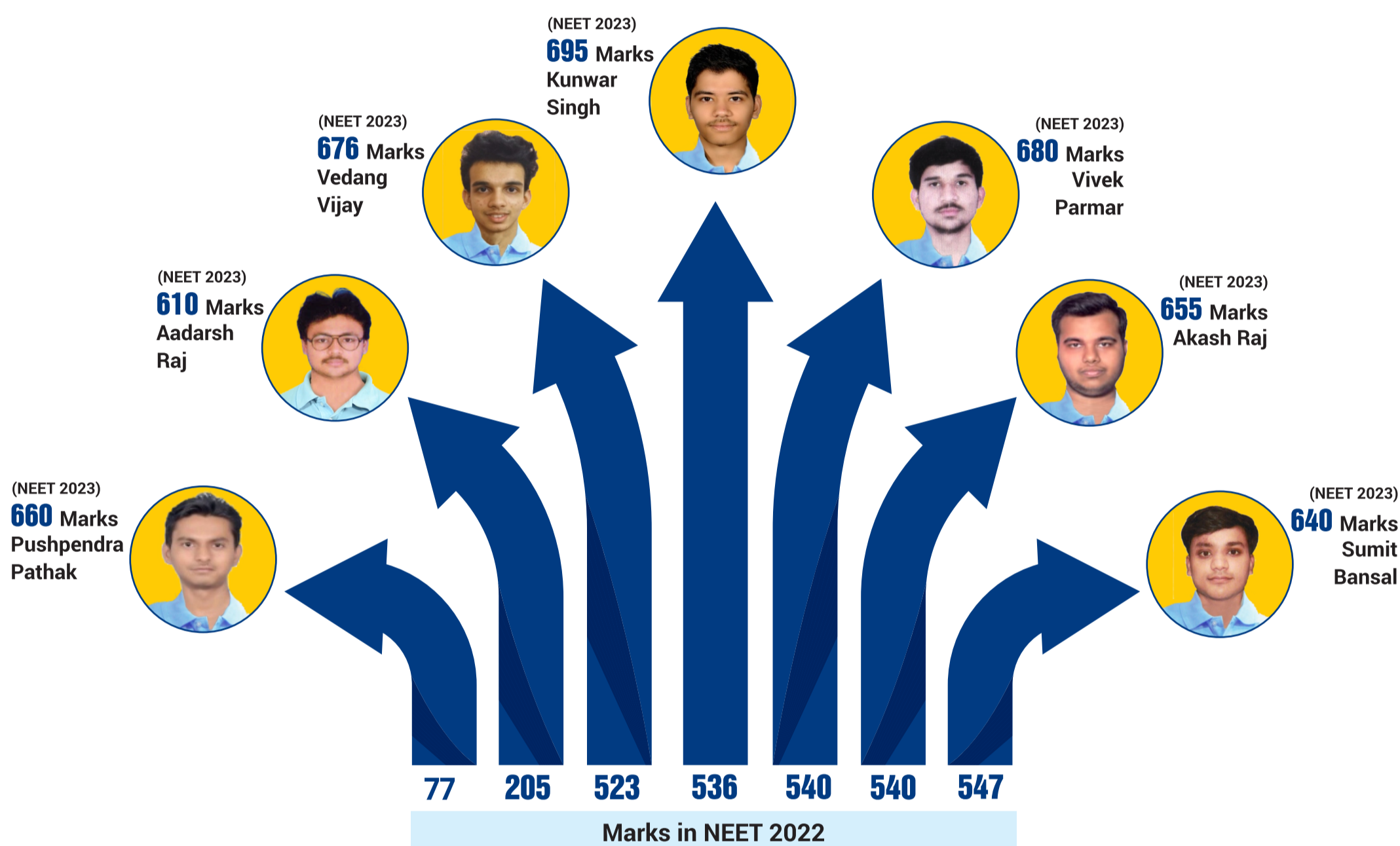
Choose the correct answer from the options given below:

- (1) A-III, B-IV, C-II, D-I
- (2) A-III, B-IV, C-I, D-II
- (3) A-I, B-III, C-II, D-IV
- (4) A-IV, B-II, C-I, D-III

Ans. [2]

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